

# Outline of the Solutions to Homework # 1

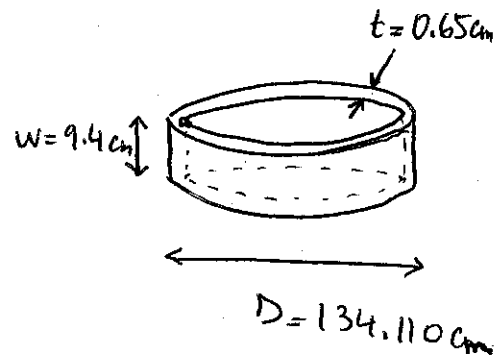
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## 17.27 (Barrel)

$$D_B = 134.122 \text{ cm at } 20^\circ\text{C}$$

$$D_{\text{iron}} = 134.110 \text{ cm at } 20^\circ\text{C}$$

$$W = 9.4 \text{ cm} \quad t = 0.65 \text{ cm}$$



$$a) \quad D_B = D_{\text{iron}} \left( 1 + \frac{\beta}{3} \Delta T \right)$$

$$\Rightarrow \Delta T = \frac{3}{\beta_{\text{iron}}} \left( \frac{D_B - D_{\text{iron}}}{D_{\text{iron}}} \right) = \frac{3}{35 \cdot 10^{-6}} \left( \frac{0.012}{134.1} \right) \approx 7.67^\circ\text{C}$$

$$T_f = 20 + 7.67 = 27.67^\circ\text{C}$$

if you add  $0.008 \text{ cm}$  to  $D_B$  :  $\Delta T = \frac{3}{35 \cdot 10^{-6}} \times \frac{0.02}{134.1} = 12.7^\circ\text{C}$

$$T_f = 20 + 12 = 32^\circ\text{C}$$

b) When it cools to  $20^\circ\text{C}$

$$\frac{\Delta D}{D} = \alpha \Delta T = \frac{\beta}{3} \Delta T = \frac{0.012}{134.1} \approx 9 \cdot 10^{-5}$$

Tension = cross section  $\times$  pressure

$$= (t \times W) E \frac{\Delta D}{D}$$

$$\text{pressure} = E \cdot \frac{\Delta D}{D}$$

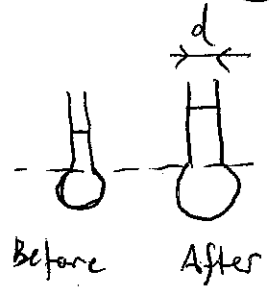
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Young Modulus of Iron

## 17.72

$\Delta V_{\text{hg}}$  of mercury is given by:  $(V_{\text{bulb}} + \frac{\pi d^2}{4} h_0) \times \beta_{\text{hg}} \Delta T$

the volume of pyrex glass is changed by  $\Delta V_{\text{pyrx}} = V_0 \beta_{\text{pyrx}} \Delta T$

Net increase in volume of mercury is  $(V_0)(\beta_{\text{hg}} - \beta_{\text{pyrx}}) \Delta T$



If the change in the bulb was neglected,

we would have  $\Delta h = h \alpha \Delta T$

with  $\alpha = \frac{1}{3} (\beta_{hg} - \beta_{pyrx})$  : coeff of linear expansion (relative) to pyrex

there is however an extra raise in height due to the expansion of Mercury in the bulb:

$$\Delta V_b = V_b (\beta_{hg} - \beta_{pyrx}) \Delta T = \frac{\pi d^2}{4} \Delta h'$$

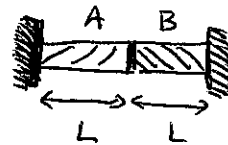
( $d$  = diameter of column)

$\Rightarrow$  The net increase in height is  $\Delta h + \Delta h' = \left( \frac{h}{3} + \frac{4V_b}{\pi d^2} \right) (\beta_{hg} - \beta_{py}) \Delta T$

If we ignore tube volume compared to bulb volume

then  $h \frac{\pi d^2}{4} \ll V_b \Rightarrow \Delta h + \Delta h' = \frac{4V_b}{\pi d^2} (\beta_{hg} - \beta_{pyrx}) \Delta T$

## Hot Rods



- After heating combined length is  $L+L=2L$  but due to their different hardness, their expansion at constant total length will be different
- Since both expand  $\rightarrow$  there will be a compressive stress in both rods. equilibrium implies it has to be the same for both rods

$$\frac{\Delta L_i}{L} = \alpha_i \Delta T \quad (\text{for each rod } i) \rightarrow \Delta L_{\text{tot}} = L (\alpha_A + \alpha_B) \Delta T$$

Just like springs put in series, the Young modulus of 2 rods

in series is  $Y = \frac{Y_A Y_B}{Y_A + Y_B} \Rightarrow \text{Total stress} = \left| \frac{F}{A} \right| = Y_{\text{tot}} \frac{\Delta L_{\text{tot}}}{L} = \left( \frac{Y_A Y_B}{Y_A + Y_B} \right) (\alpha_A + \alpha_B) \Delta T$