

## Lecture 3

### April 6, 2012

So far we discussed various types of energy, now we take a simple illustrative calculation.

Problem:

Calculate the mass energy available in a single hydrogen atom expressed in Joules and MeV

•Solution.

•Idea: calculate the mass in kg of the Hydrogen atom and use Einstein's eqn  $E = MC^2$

•Mass = 1 proton + 1 electron = 1 proton (electron is lighter by factor of 1000) =  $1.67 \times 10^{-27}$  kg

• $C = 3 \times 10^8$  meters per second

• $E = 1.503 \times 10^{-6}$  Joule = 939 MeV

Similarly a single electron has mass energy = .5 MeV

Energy equivalents				
<i>Conversion table</i>				
	J	kWh	Btu	
1 Joule	1	$2.78 \times 10^{-7}$	$9.49 \times 10^{-4}$	
1 kWh Kilowatt Hour	$3.60 \times 10^6$	1	3413	
1 calorie	4.184	$1.16 \times 10^{-6}$	$3.97 \times 10^{-3}$	
1 British Thermal Unit BTU	1055	$2.93 \times 10^{-4}$	1	
1 ft pound (ft-lb)	1.36	$3.78 \times 10^{-7}$	$1.29 \times 10^{-3}$	
1 electron volt (eV)	$1.60 \times 10^{-19}$	$4.45 \times 10^{-26}$	$1.52 \times 10^{-22}$	
1 Barrel petroleum (42 US Gallon)	$6.12 \times 10^9$	1700	$5.8 \times 10^6$	

In this way we can find the energy equivalent of any mass. In using this formula, we will usually first have to find the mass defect in some reaction, i.e. the difference in mass between LHS and RHS and then multiply by  $C^2$  to get the energy.



# Quick review of energy/power

# Some real life example

$$\text{Work} = \text{Force} \times \text{Length}$$

Force

$$\text{Energy} = \text{Power} \times \text{time}$$

$$\text{Energy} = \text{Work done}$$

$$\text{Power} = \text{Rate of doing work}$$

MKS

Newton

e.g. 1 KG weight suspended in gravity by a string exerts a force 9.8 Newtons

Joule = Newton Meter

e.g.

20 KG suitcase pulled up by 10 meters costs ~1960 Joules

Watt = Joule/sec

e.g. if the suitcase is hauled up in 3.25 mins (~190 sec) power used is 10 Watts!

FPS

Pound (lb)

Foot pound (ft lb)

1 Joule = 1.32 ft lb

The FPS system can be very confusing since pound is used in many contexts: Our strategy is to convert ft-lb to Joule as fast as we can!!  
In above example the work done is ~2500 J = 2.5 KJ

Horse power = 550 ft lb/sec

Convert to Watts using

1 HP = 746 Watts

In above problem the "puny human" uses a horse power ~.013



## Start thinking of energy content

**Table 1.3: Energy Content of Fuels**

Type of Fuel	Energy in joules/kg	Type of Fuel	Energy in joules/kg
Coal	$2.9 \times 10^7$	Garbage and Trash	$1.2 \times 10^7$
Crude Oil	$4.3 \times 10^7$	Bread	$1.0 \times 10^7$
Gasoline	$4.4 \times 10^7$	Butter	$3.3 \times 10^7$
Natural Gas	$5.5 \times 10^7$	Nuclear fission with Uranium 235	$8.0 \times 10^{13} = 8,000,000 \times 10^7$
Wood	$1.4 \times 10^7$		



From this table we see that Uranium has huge amount of energy: as compared to converting garbage and trash.

1kg Uranium is appxly 6.7 Million kg of trash!!!

In practice the energy contained in nuclear reactions far exceeds any other source. However the risks involved are high, as seen in recent Fukushima incident and before that in Chernobyl and the Three mile island.

We will discuss this aspect later.

Most materials give us  $\sim 10^7$  J/kg, not Uranium



# Stepping up to bigger scales

Numbers useful on a bigger scale				
Total energy consumption in USA				
(1 QBtu ~ referred to as a Quad)		1 Quad=172.4 Million Barrels		
1 Quad = 293 TWh		1000TWh=3.41 Quad		
2003	98.3 Quad	$2.88 \times 10^{16}$	watt hours	28800 TWh
2007	101.6 Quad	$2.97 \times 10^{16}$	watt hours	29700 TWh
Annual electricity production in USA in 2010				
3992 TWh		(China 3715 TWh)		
Energy Consumption per capita in US = $101.6 \times 172.4 \text{ Mill} / 290 \text{ Mill} = 60.4$ barrels per person				

1 Tera Watt Hour =  $10^{15}$  Watt Hour = Billion KWH

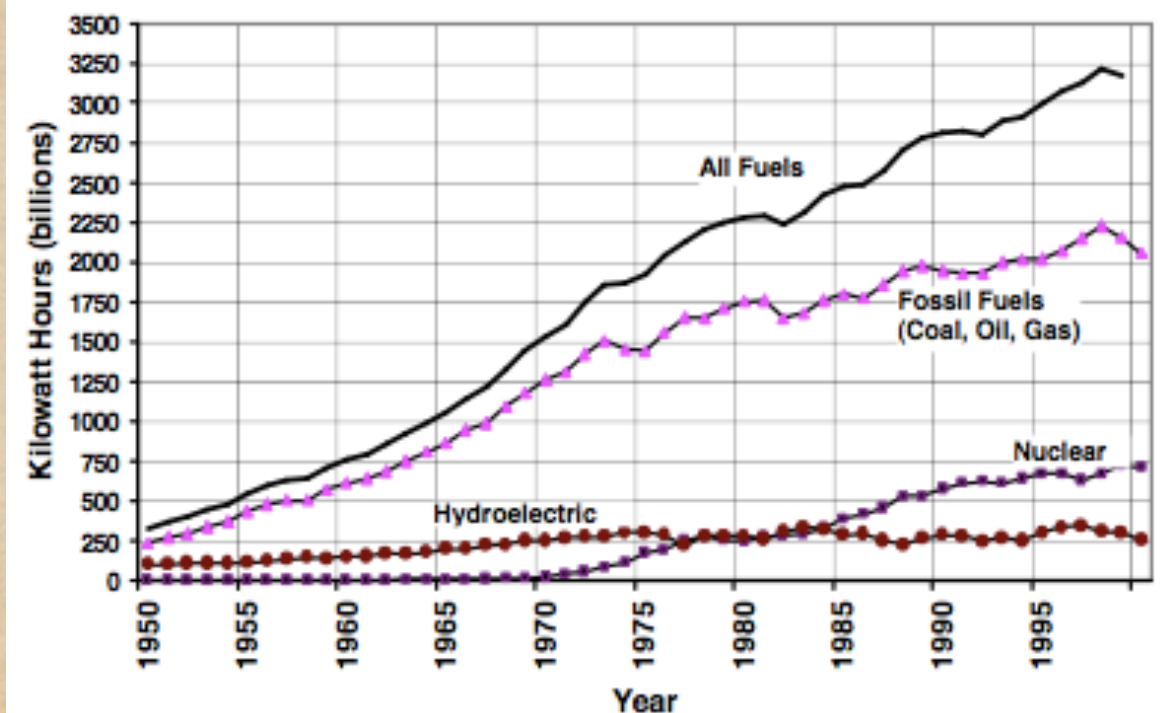
(since Billion =  $10^9$ )

Watt Hour = 3600 Joules

Therefore TWh =  $3.6 \times 10^{18}$  Joules

US production ~  $1.4 \times 10^{23}$  Joules

Fig. 1.1: Annual Electricity Production in the U.S. by Type of Fuel





Total energy consumption in USA				
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Energy Consumption per capita in US = 101.6x 172.4 Mill/290 Mill =  
60.4 barrels per person

300 Giga Joule/ person /year







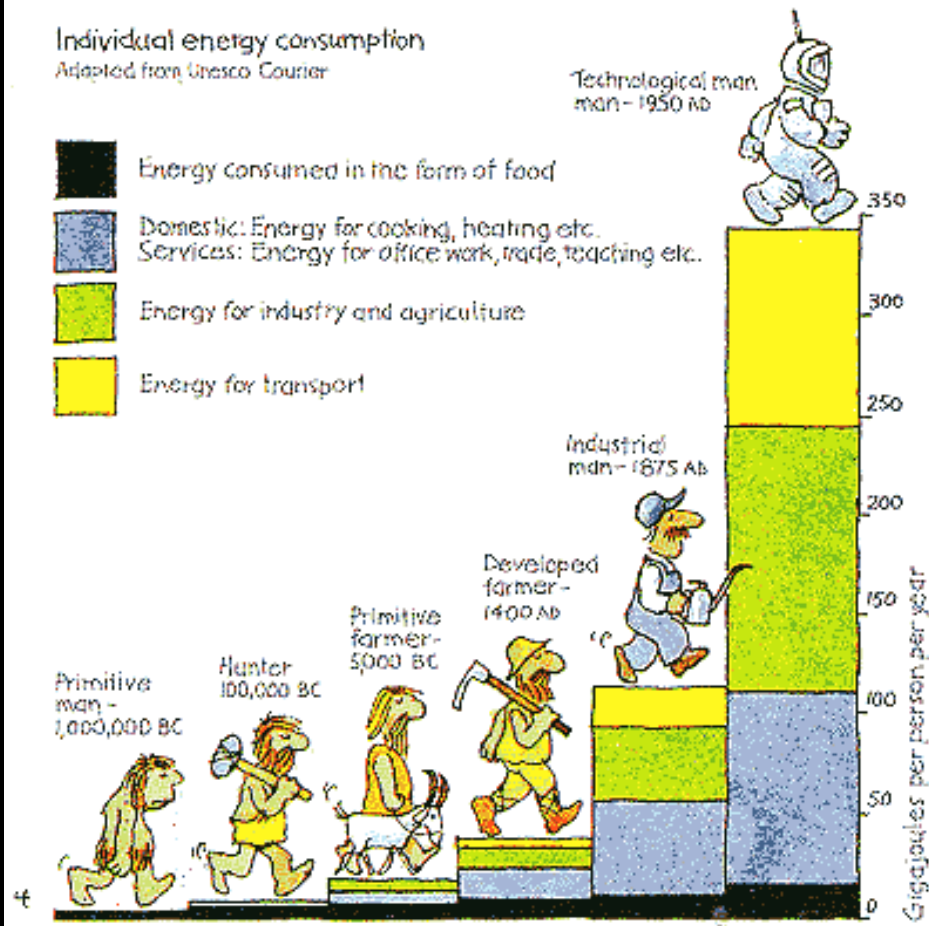


1 GigaJoule =  $10^9$  J  
 6.12 GigaJoule = 1 Barrel  
 350 GigaJoule = 57 Barrels

Individual energy consumption

Adapted from Unesco Courier

-  Energy consumed in the form of food
-  Domestic: Energy for cooking, heating etc.  
Services: Energy for office work, trade, teaching etc.
-  Energy for industry and agriculture
-  Energy for transport



Today 28% of the world's population consumes 77% of the world's energy production.

Or 3/4 of the world's population uses less than 1/4 of the energy produced



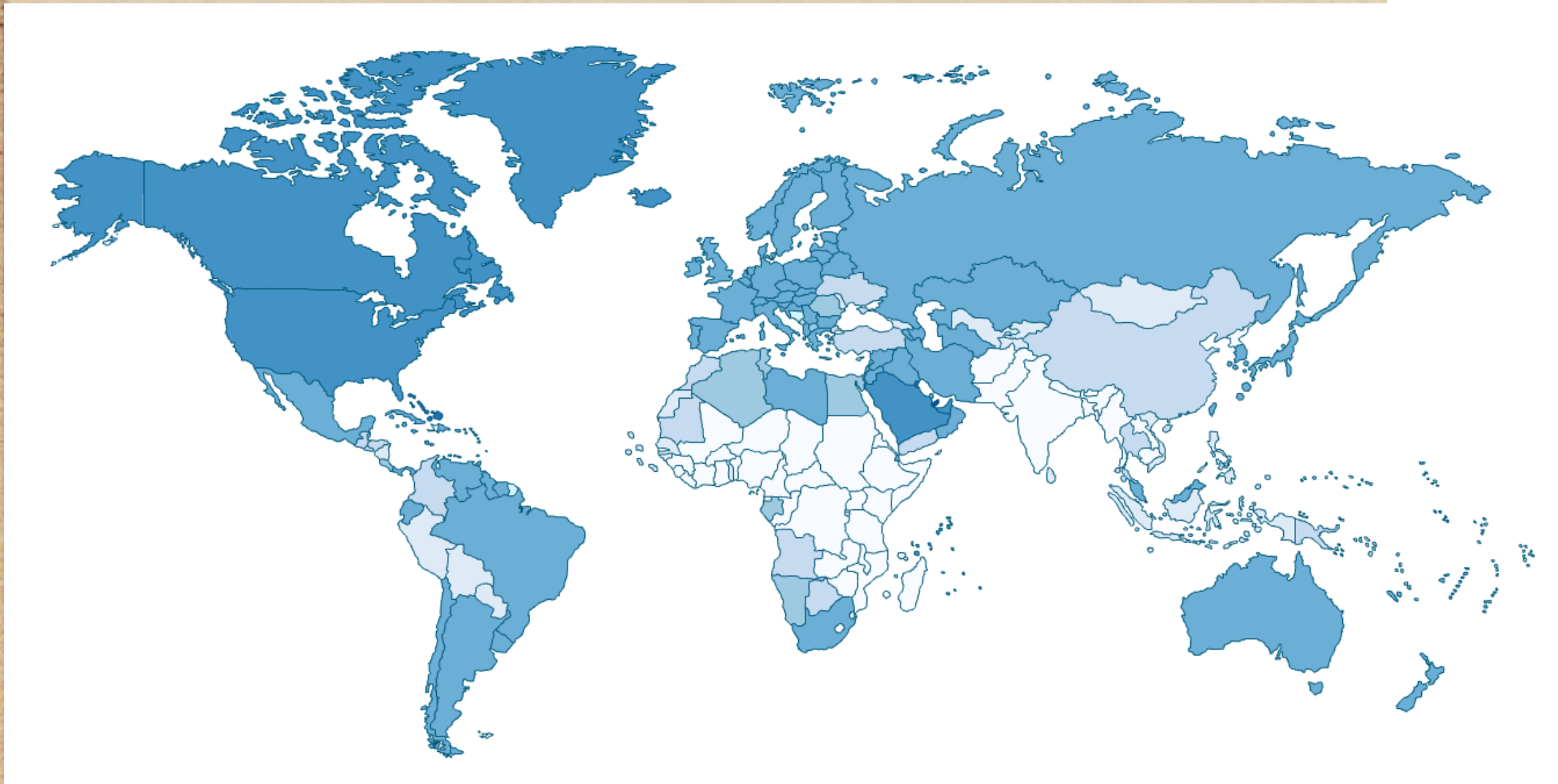
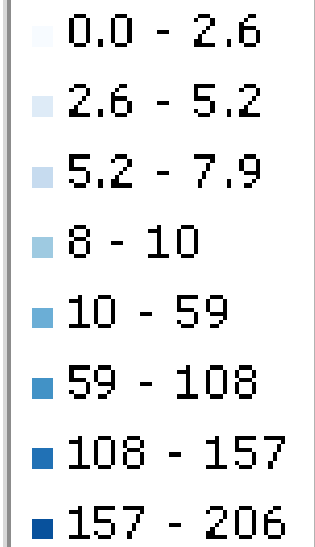
# Individual Energy Consumption across the Ages

Today, a man uses **100 times** the Energy his primitive ancestors used to consume!



# Oil consumption per capita (bbl/day per 1000 people)

( multiply by .36 to get barrels per person per year)





## Big question and thoughts

- Why do we use so much energy?
- Poor efficiency in part
- “Unconscious” Habits of affluent civilizations
- Poor or exuberantly optimistic planning
- Will we have to change our energy usage patterns?

Fossil fuel is a single time usage affair and is far from being inexhaustible.

M King Hubbert's realization.

Today we distinguish between

Renewable and non renewable fuel.

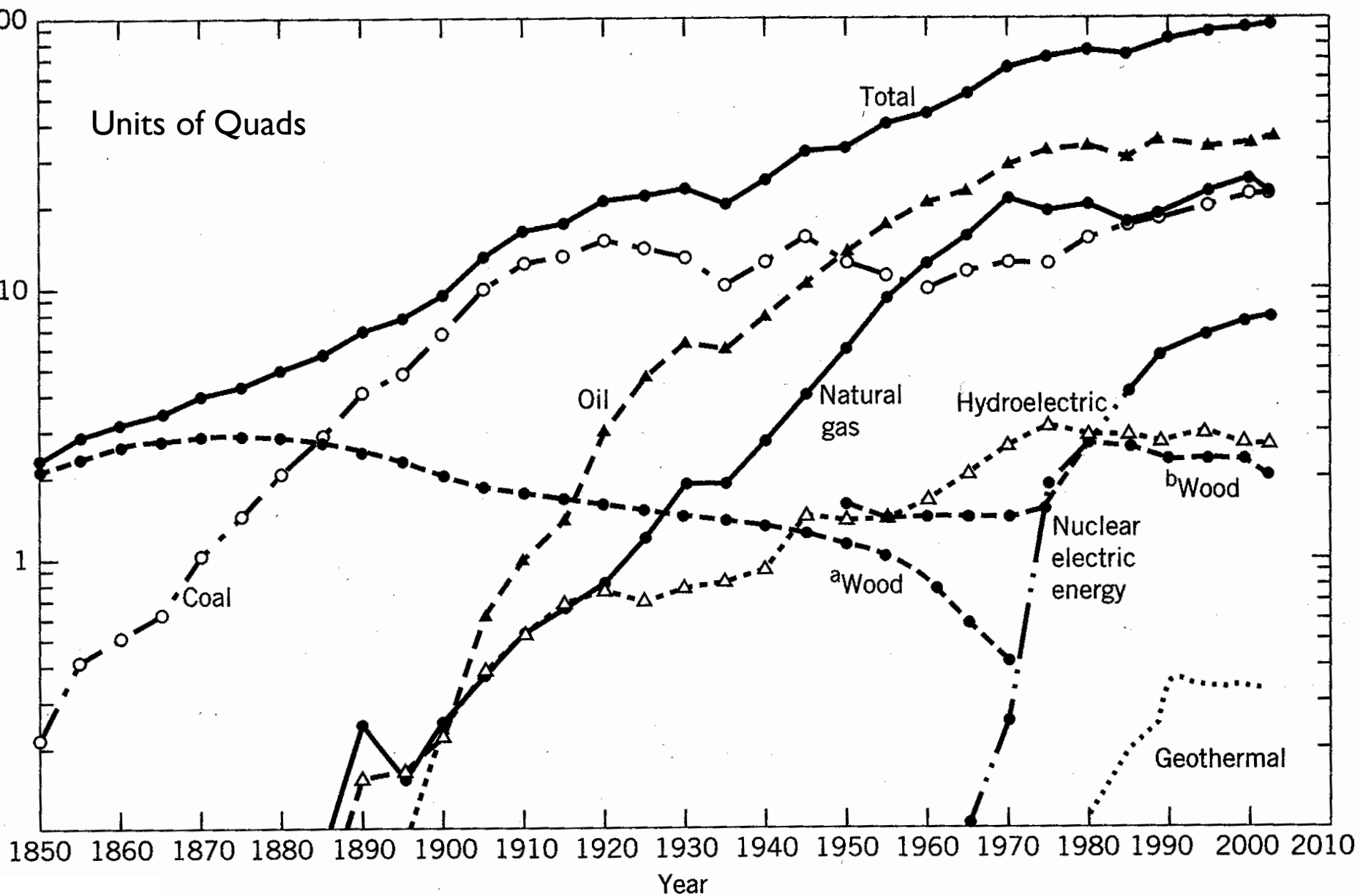
Renewable implies that nature replenishes the resource over time.

Non renewable is clearly everything else.



Different forms of energy in the US

Current table on web in resources folder



Various forms of energy consumed in the United States since 1850. This

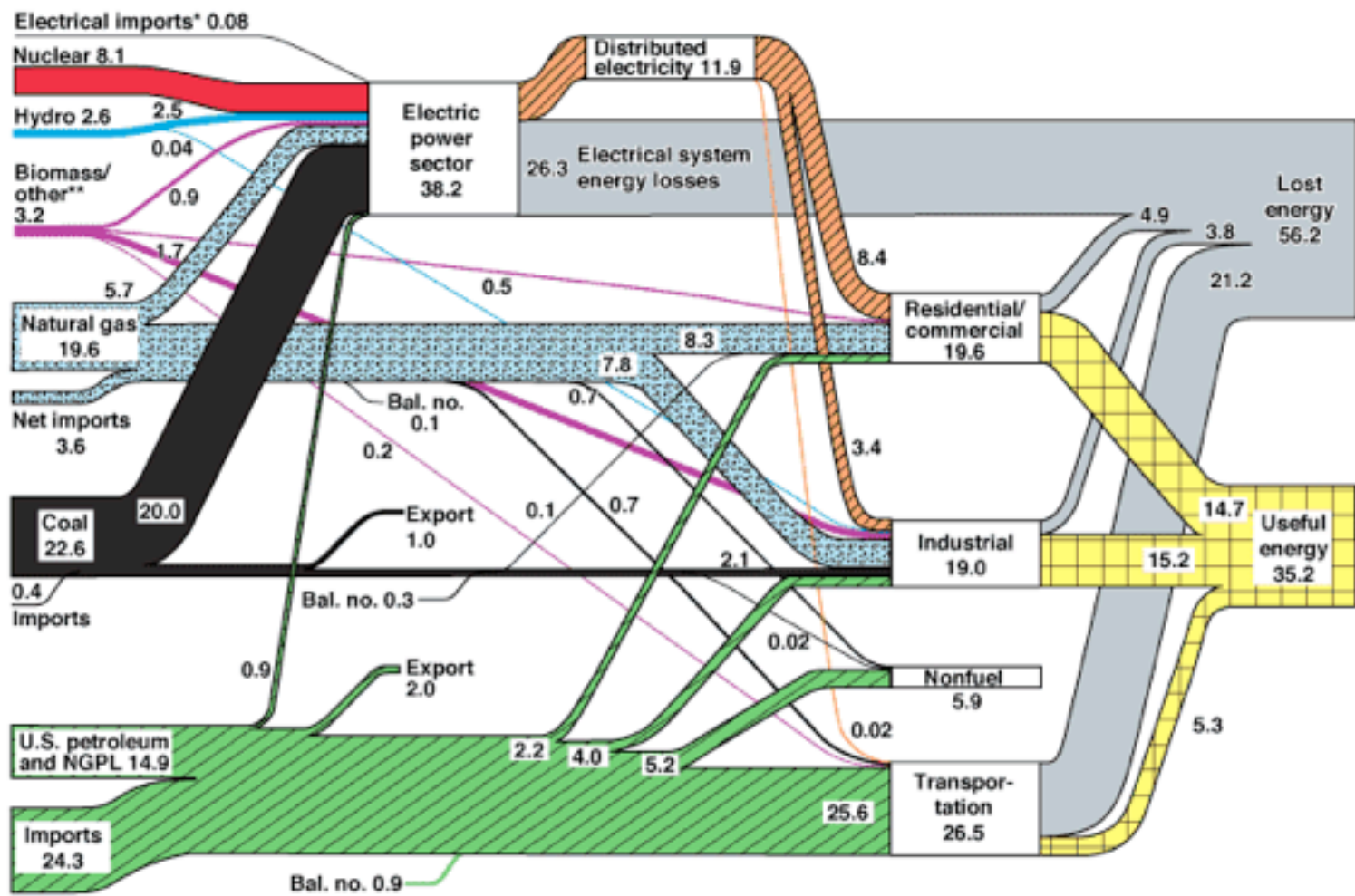
- Renewable energy
- ◆ Biofuel
- ◆ Biomass
- ◆ Hydroelectricity
- ◆ Solar energy
- ◆ Tidal power
- ◆ Wave power
- ◆ Wind power



# Energy equation Expense vs. Sources of energy

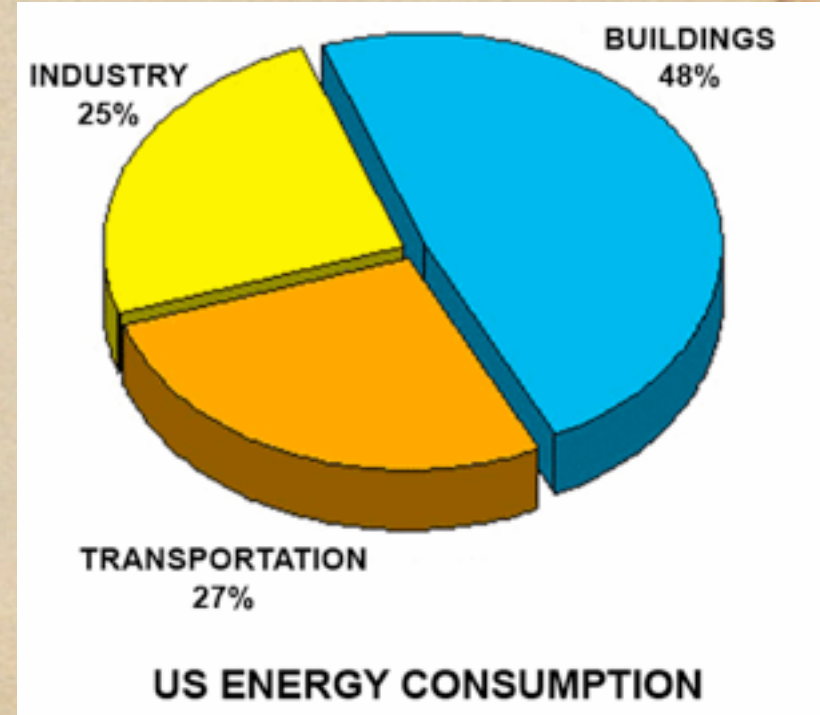
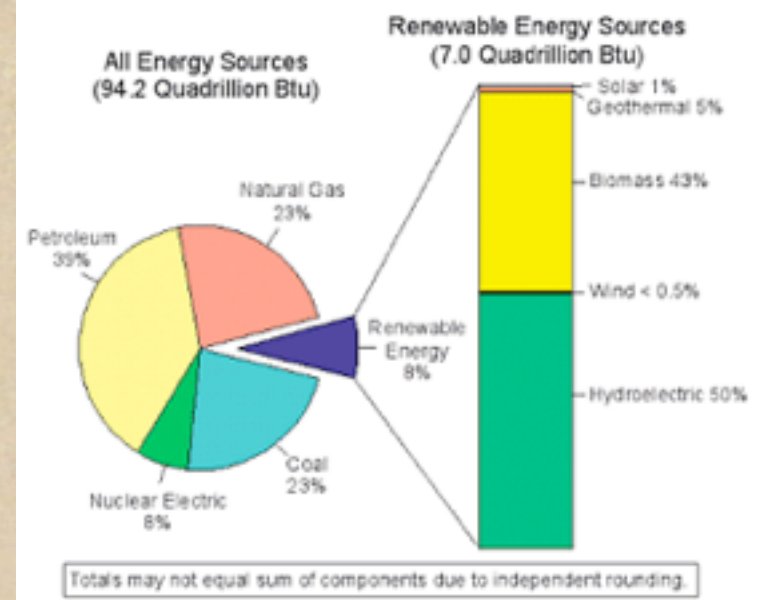
Flow chart of energy eqn

## U.S. Energy Flow Trends – 2002 Net Primary Resource Consumption ~97 Quads



Source: Production and end-use data from Energy Information Administration, Annual Energy Review 2002.  
\*Net fossil-fuel electrical imports.  
\*\*Biomass/other includes wood, waste, alcohol, geothermal, solar, and wind.

June 2004  
Lawrence Livermore  
National Laboratory  
<http://eed.llnl.gov/flow>





Renewable energy is energy which comes from natural resources such as sunlight, wind, rain, tides, and geothermal heat, which are renewable (naturally replenished). In 2008, about 19% of global final energy consumption came from renewables,

The share of renewables in electricity generation is around 18%, with 15% of global electricity coming from hydroelectricity and 3% from new renewables.

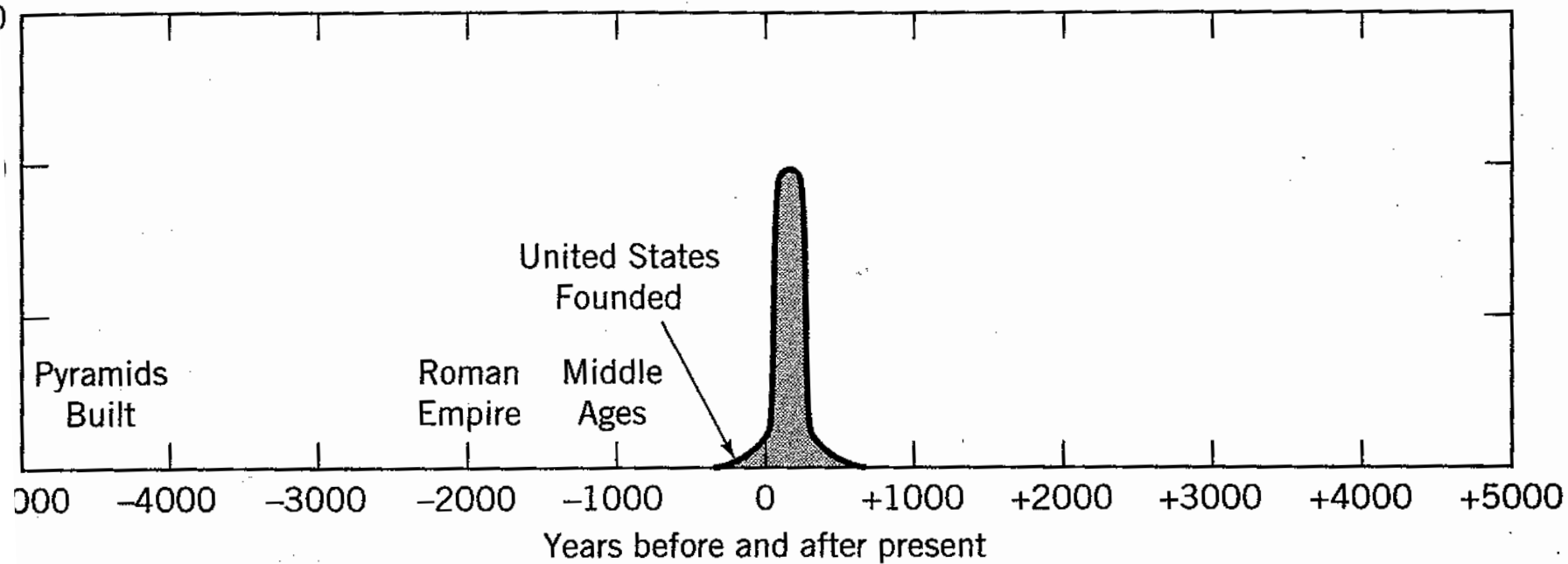
Wind power is growing at the rate of 30% annually, with a worldwide installed capacity of 158 gigawatts (GW) in 2009, and is widely used in Europe, Asia, and the United States. At the end of 2009, cumulative global photovoltaic (PV) installations surpassed 21 GW and PV power stations are popular in Germany and Spain. Solar thermal power stations operate in the USA and Spain, and the largest of these is the 354 megawatt (MW) SEGS power plant in the Mojave Desert. The world's largest geothermal power installation is The Geysers in California, with a rated capacity of 750 MW. Brazil has one of the largest renewable energy programs in the world, involving production of ethanol fuel from sugar cane, and ethanol now provides 18% of the country's automotive fuel. Ethanol fuel is also widely available in the USA.

The **DeSoto Next Generation Solar Energy Center** is a [photovoltaic solar power](#) facility in [Arcadia, DeSoto County, Florida](#) owned by [Florida Power & Light \(FPL\)](#)



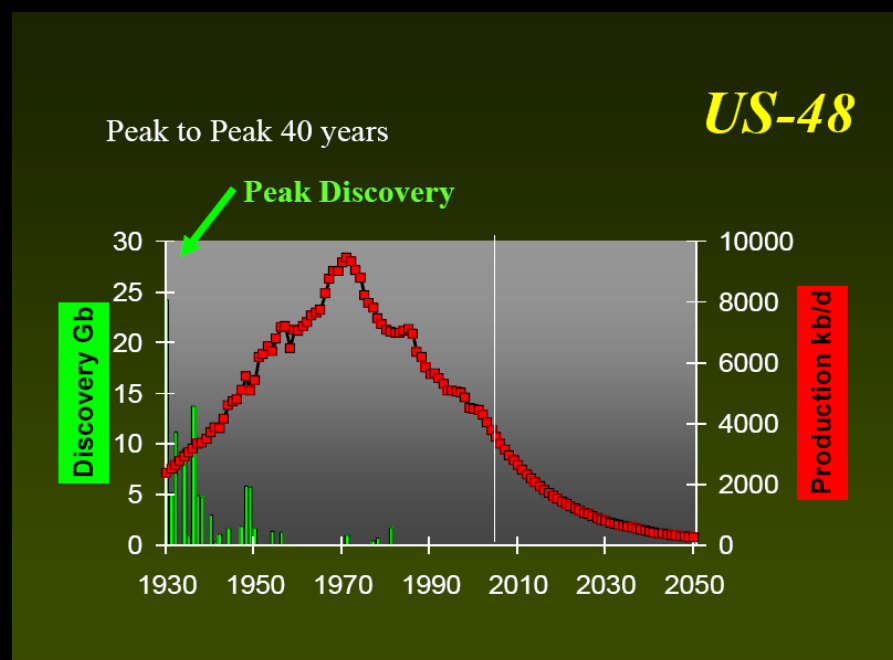


## Goodbye Fossil Fuels!



**1.2** The complete exploitation of the world's fossil fuels will span only a relatively brief time in the 10,000 year period shown centered around the present. (Source: cited with permission from M. K. Hubbert, *Resources and Man*, Washington, National Academy of Sciences, 1969. Historical events added.)

### The now famous plot of US-48



M King Hubbert predicted in a famous report written in 1956 (available on the resource page) that the peak of discovery of oil fields in 1932 followed by a drop off, would be mirrored in the oil production curve with a time lag caused by the development activity.



Some problems relating to conversions  
with useful hints

1. If you push a cart along a horizontal surface with a force of 10 pounds, and the cart moves 10 feet, how much work have you done in ft·lbs? In joules?

Hint: (1) Identify the formula needed (2) Plug in the numbers.

$$W = F \times L$$

$$10 \times 10 = 100 \text{ FootPounds} = 136 \text{ Joules}$$

2. How many tons of coal would be needed each year to provide for the entire energy needs for the average person in the United States?

Hint: This problem requires us to find out the energy content in coal by looking up a table, finding out how much energy is needed per person in USA from another table and combining the two.

$$60 \text{ barrels per person} = 60 \times 6.12/42 \times 10^9 = 8.7 \times 10^9 \text{ Joules}$$

$$1 \text{ Ton coal} = 2.8 \times 10^{10} \text{ Joules}$$

Answer 0.31 Tons coal per person per year!!



3. Solar energy is incident on a black parking lot with an intensity of  $1000 \text{ Watts/m}^2$  and 90% of it is absorbed. What is this in Btu/hr per square meter? What happens to the other 10%?

Hint: This problem is much simpler than it looks at first sight. We can take 1 square meter as the relevant area, and then we just need to convert from watts to BTU/hour

Table tells us that 1 Kilowatt hour = 3413 Btu

Hence answer =  $.9 \times 3413 = 3071 \text{ Btu/hr}$

4. Palo Verde Power Plant, is a nuclear power plant located in Wintersburg, Arizona, about 45 miles west of central Phoenix. It is currently the largest nuclear generation facility in the United States, averaging over 3.2 Gigawatts (GW) of electrical power production in 2003.

Power plants reckon production by GigaWatt

1 year has  $3.15 \times 10^7$  secs  
or  
8760 hours.

Note the units:

1 gigawatt =  $10^9$  watts

Tera =  $10^{12}$

Total power calculation:

This figure gives us the power available on average through 2003.

If we want the total power during the year 2003, we multiply by  $3.15 \times 10^7$  secs (the number of seconds per year)

to get  $1.008 \times 10^{17}$  Joule

This is better expressed in terms of Watt hours:

(1kW Hour =  $3.6 \times 10^6 \text{ J}$ ) as

$2.8 \times 10^{13}$  WattHours = 28 TWH

Recall that the total electrical power generation in the US in 2010 is 3992 TWh