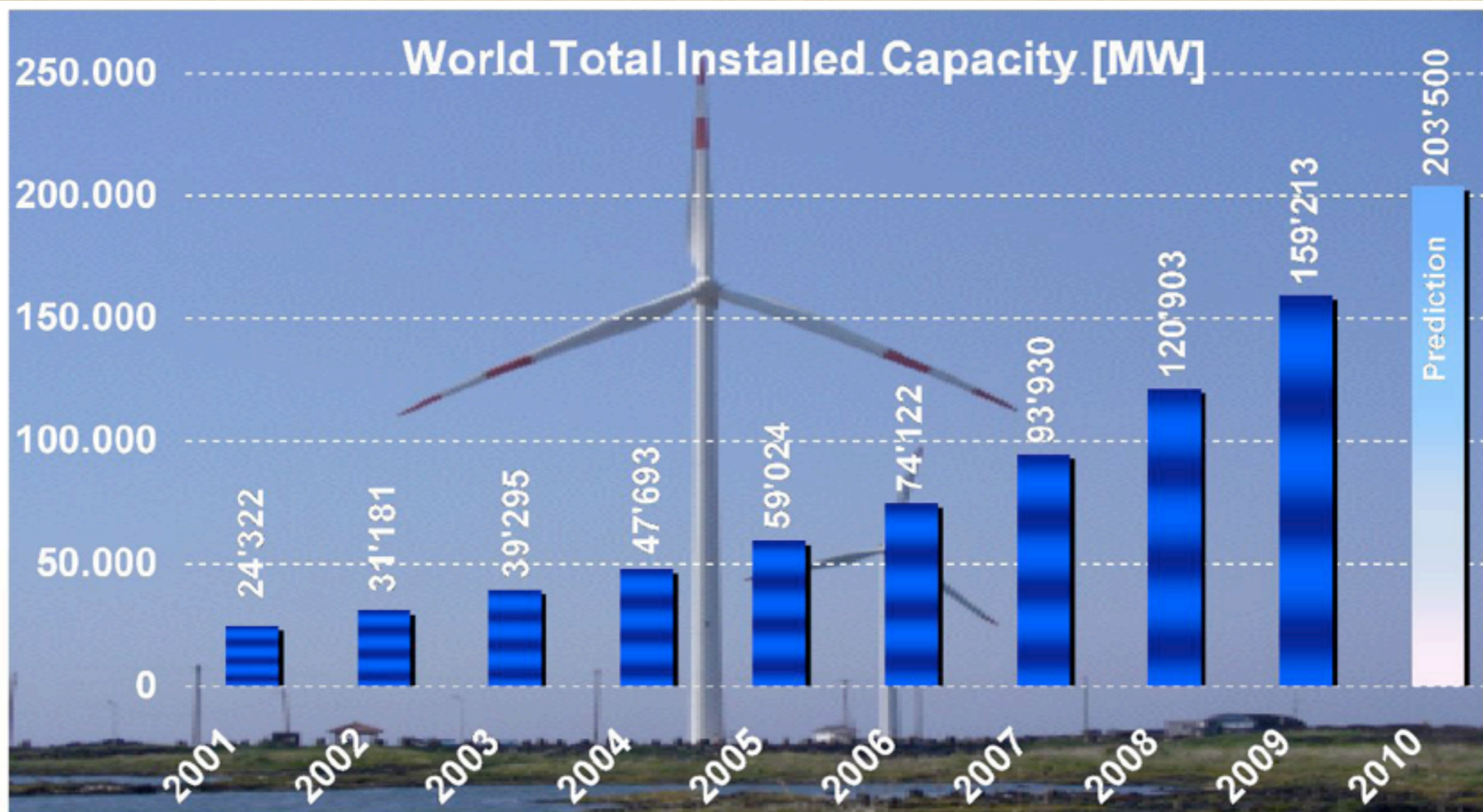


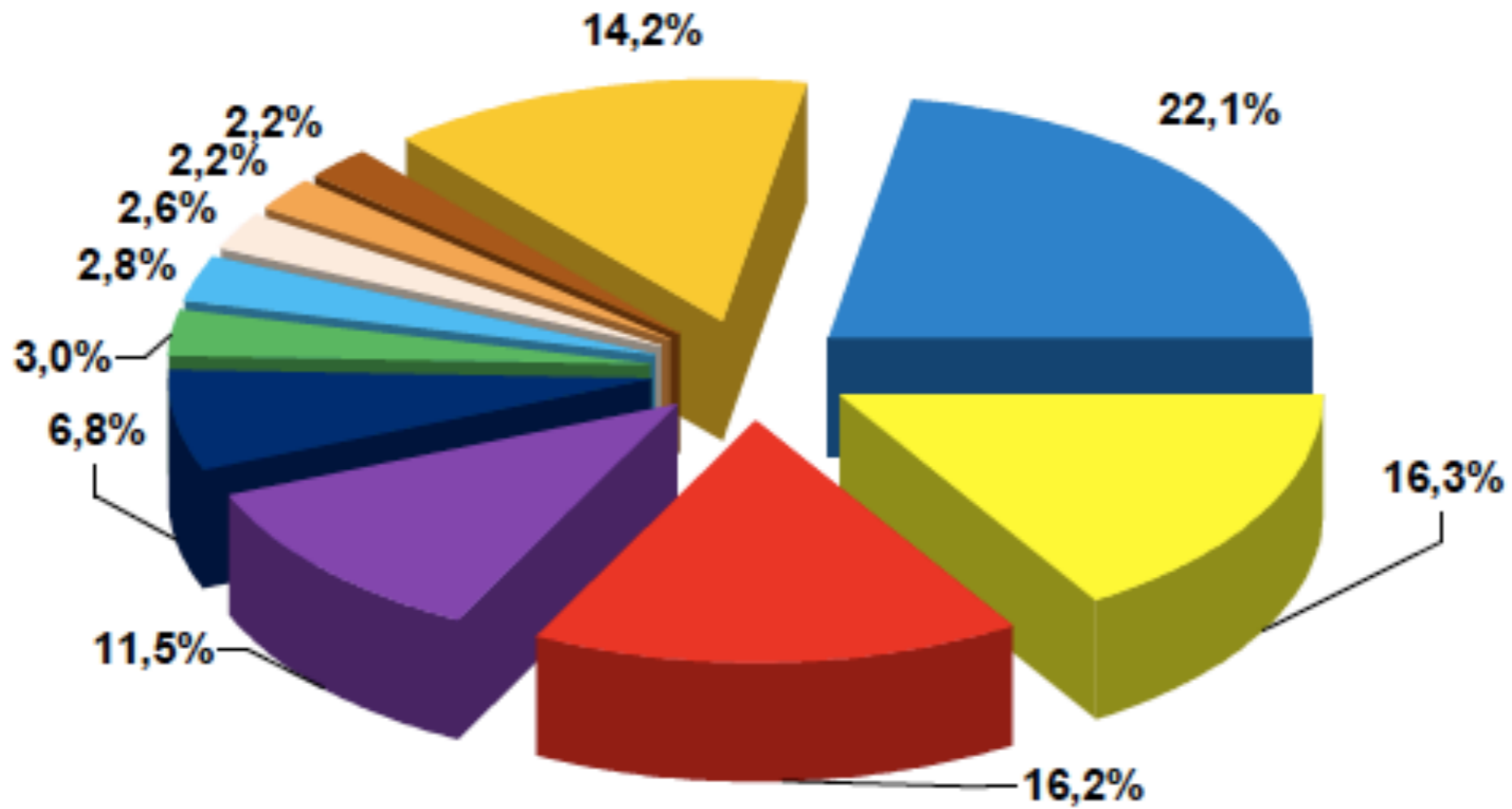
Lecture 16 May 24, 2011
Statistics of Wind power



- The trend continued that wind capacity doubles every three years.
- All wind turbines installed by the end of 2009 worldwide are generating 340 TWh per annum, equivalent to the total electricity demand of Italy, the seventh largest economy of the world, and equalling 2 % of global electricity consumption.

$159213 \text{ MW} \times 365 \times 24 = 1395 \text{ TWh}$. Clearly we are using only 1/3rd for operational reasons.

Country Share of Total Capacity 2009



- USA
- Spain
- France
- Denmark
- China
- India
- United Kingdom
- Rest of world
- Germany
- Italy
- Portugal

Prospects for a World Powered Predominately by Solar and Wind Energy

Walter Kohn
2011

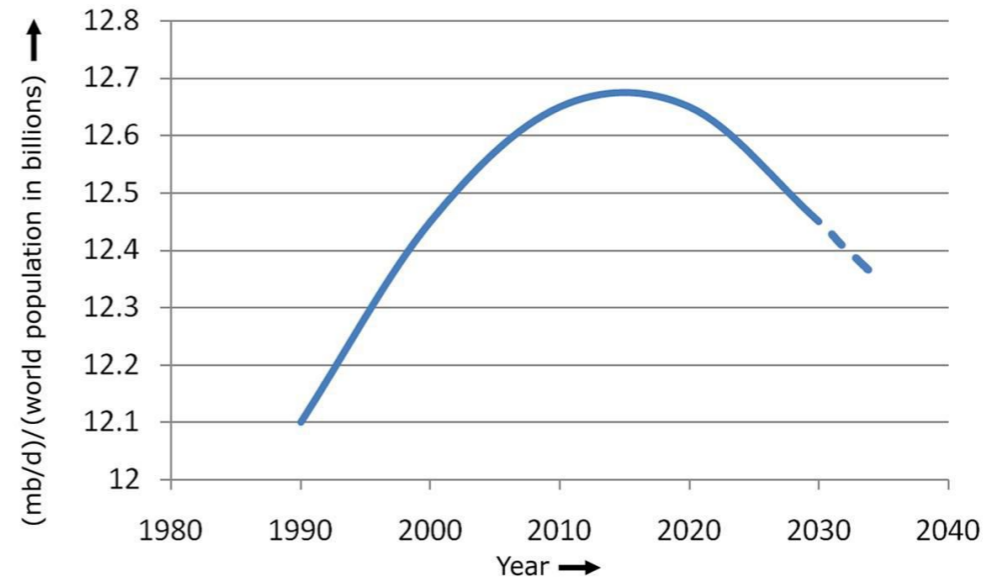
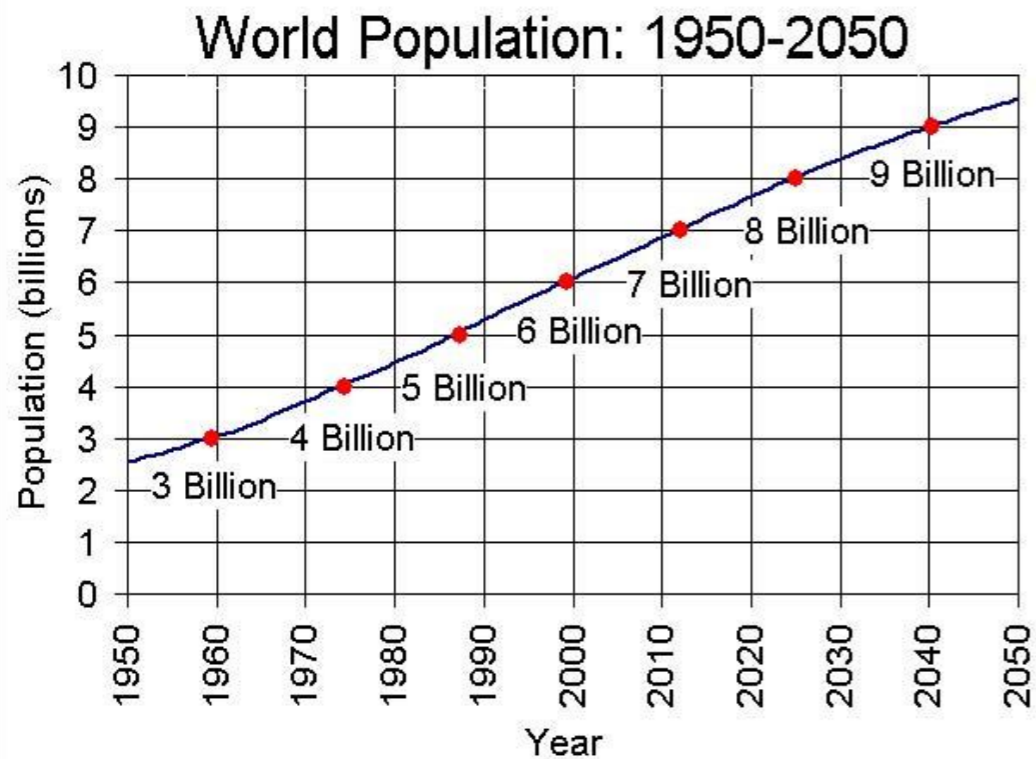


Fig. 4. Global oil production per person



Source: U.S. Census Bureau, International Data Base, December 2008 Update.



"We have met the enemy and he is us"

Pogo quote of 1971

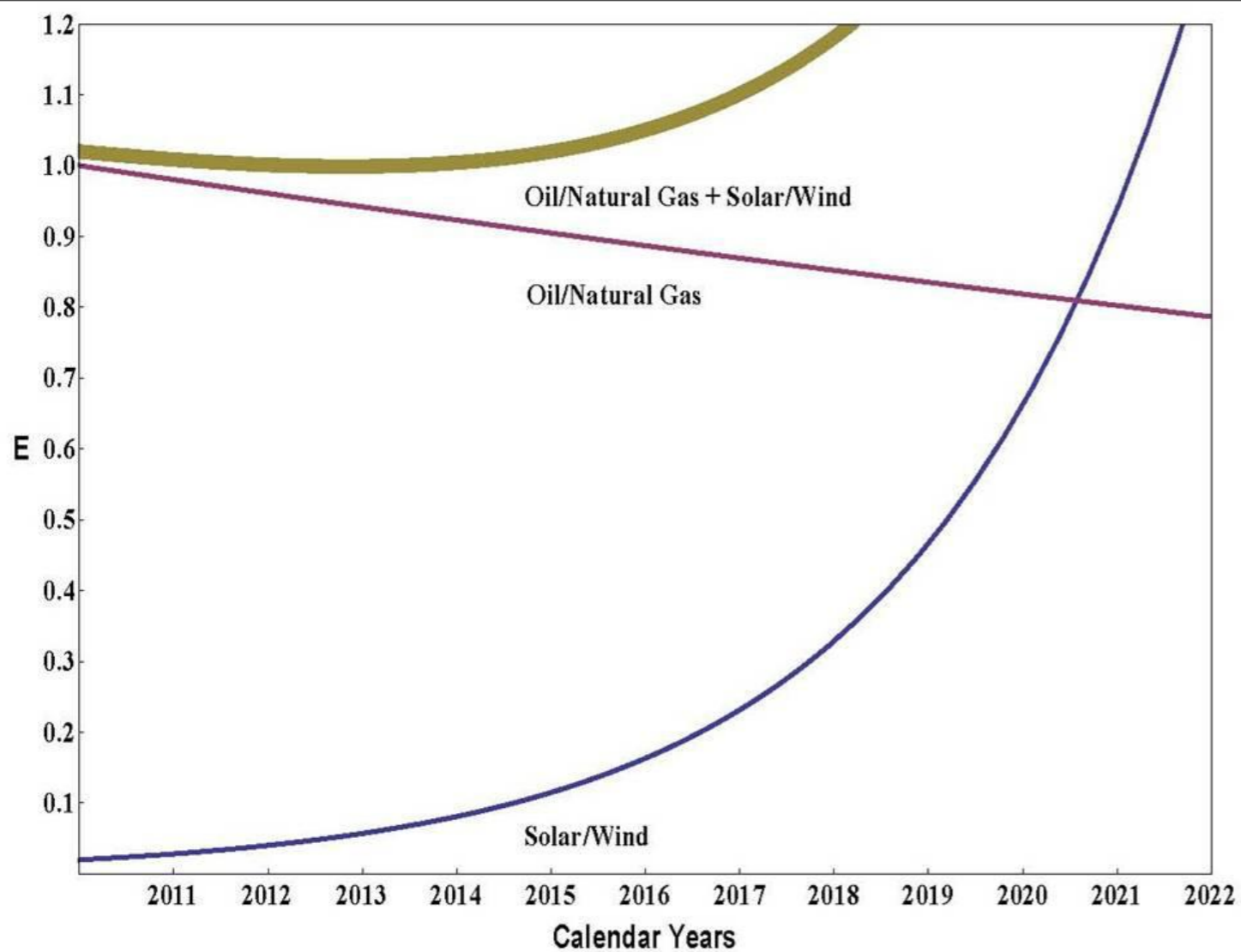
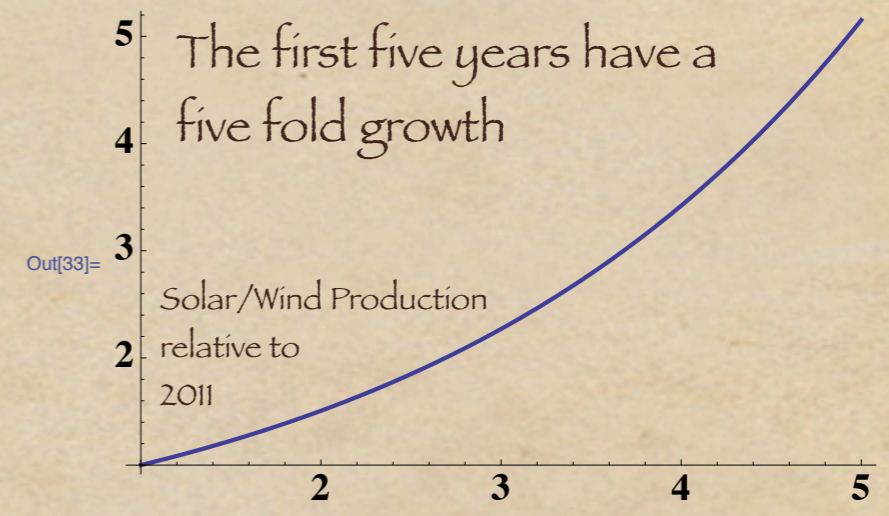
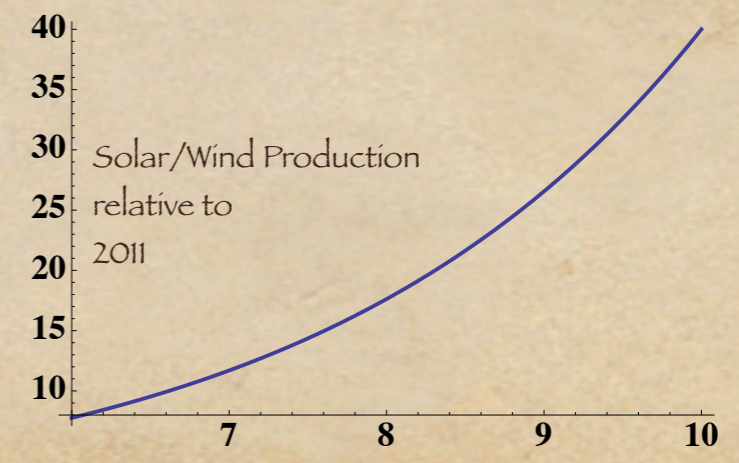


Fig. 5 The energy transition from (Oil/Natural Gas) to (Solar/Wind). We define the transition year as the year (2021) in which solar/wind energy begins to exceed oil/natural gas energy, and becomes the world's dominant energy source. E represents annual rates of energy production, in units of oil/gas production in 2010.

- Comments:
- 1) Kohn's prediction is that 2020 will see a transition to Solar/Wind domination
 - 2) He is predicting a 100 fold growth in the next 10 years!!
 - 3) Can technology fulfil this Kohn's law?
 - 4) More important perhaps than Moore's law in silicon valley.

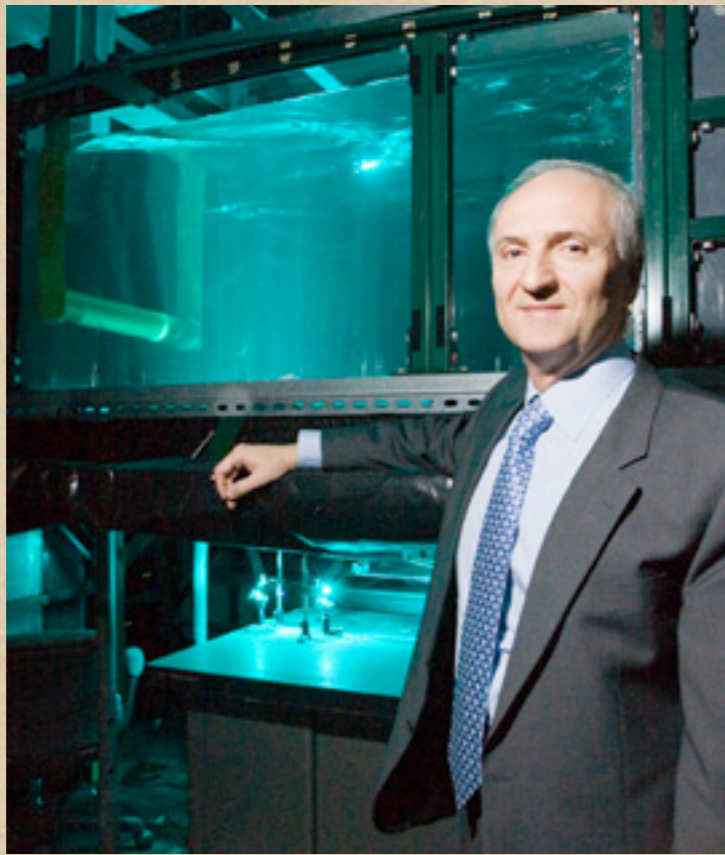


The big catchup though happens after year 6 assuming continuing exponential growth!



<http://www.vortexhydroenergy.com/>

Energy from waves!!!

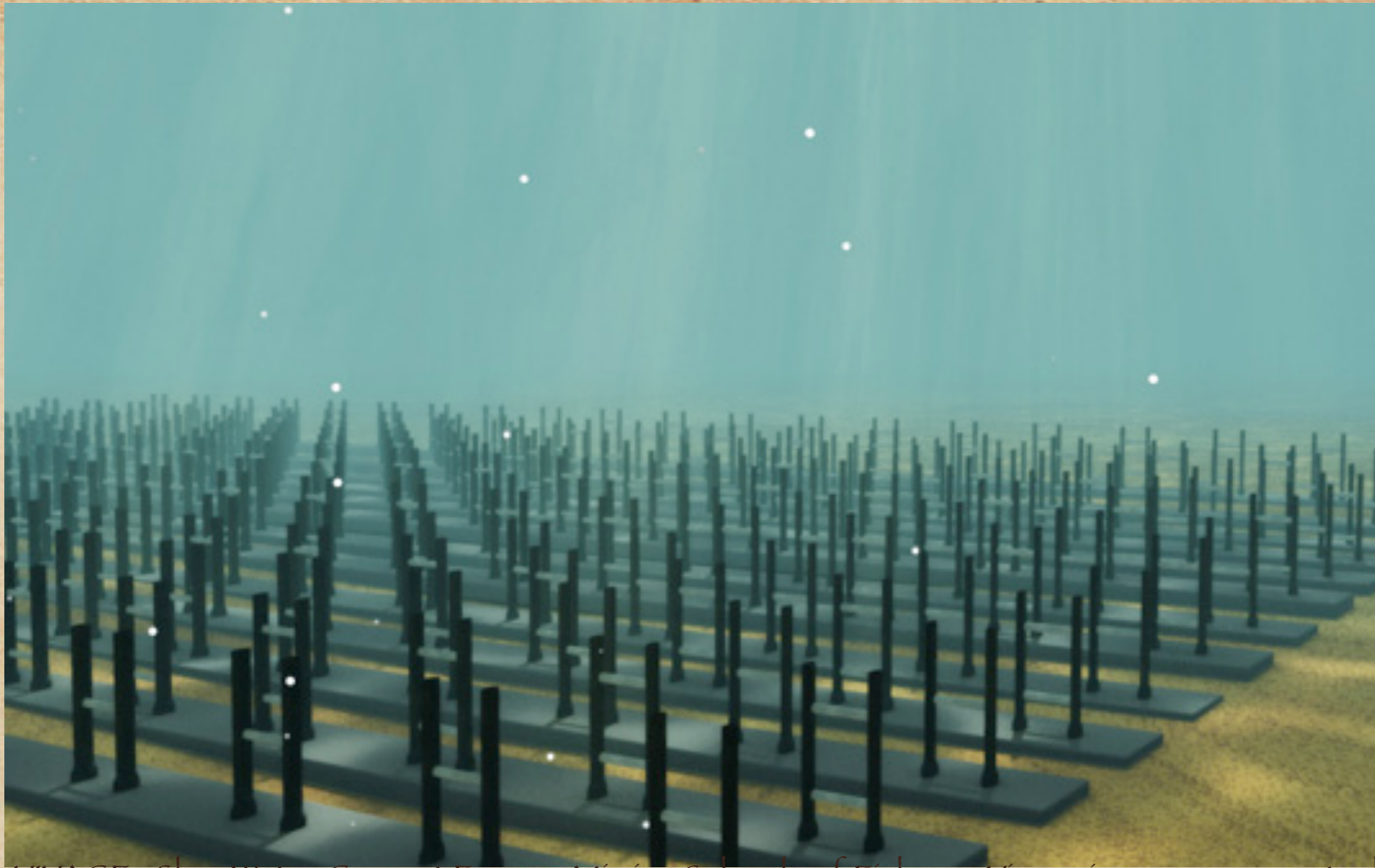


Slow-moving ocean and river currents could be a new, reliable and affordable alternative energy source. A University of Michigan engineer has made a machine that works like a fish to turn potentially destructive vibrations in fluid flows into clean, renewable power.

The machine is called VIVACE. A paper on it is published in the current issue of the quarterly *Journal of Offshore Mechanics and Arctic Engineering*

VIVACE is the first known device that could harness energy from most of the water currents around the globe because it works in flows moving slower than 2 knots (about 2 miles per hour.) Most of the Earth's currents are slower than 3 knots. Turbines and water mills need an average of 5 or 6 knots to operate efficiently.

VIVACE stands for Vortex Induced Vibrations for Aquatic Clean Energy. It doesn't depend on waves, tides, turbines or dams. It's a unique hydrokinetic energy system that relies on "vortex induced vibrations."



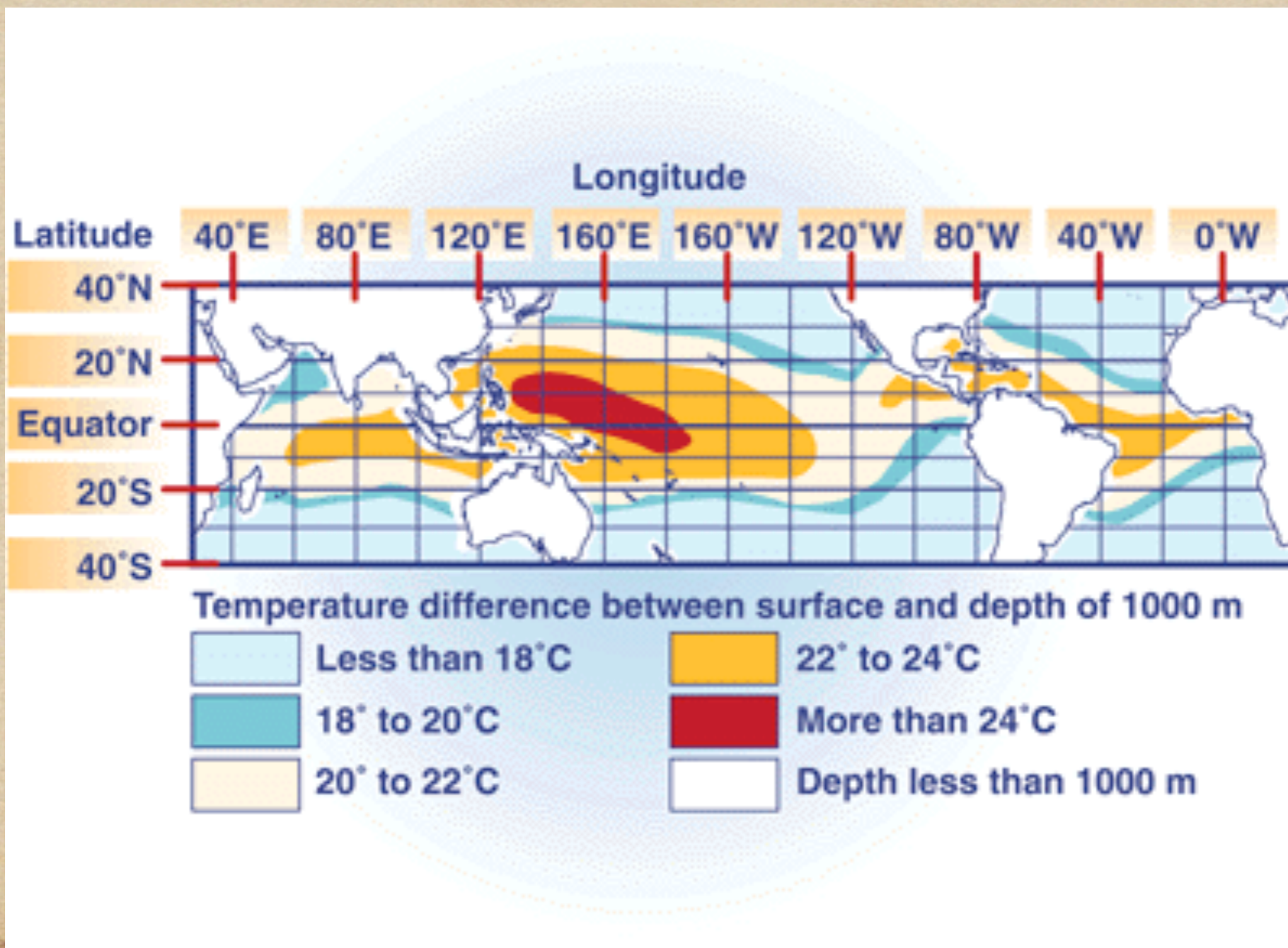
VIVACE: Slow Water Current Energy Mimics Schools of Fish: Vivace is a new energy technology that gets its name from a phenomenon that engineers have been battling for 25 years.

VIV (vortex induced vibrations) destroyed the Narrows Bridge in Washington State in 1940, and the Ferrybridge power station cooling towers in England in 1965.

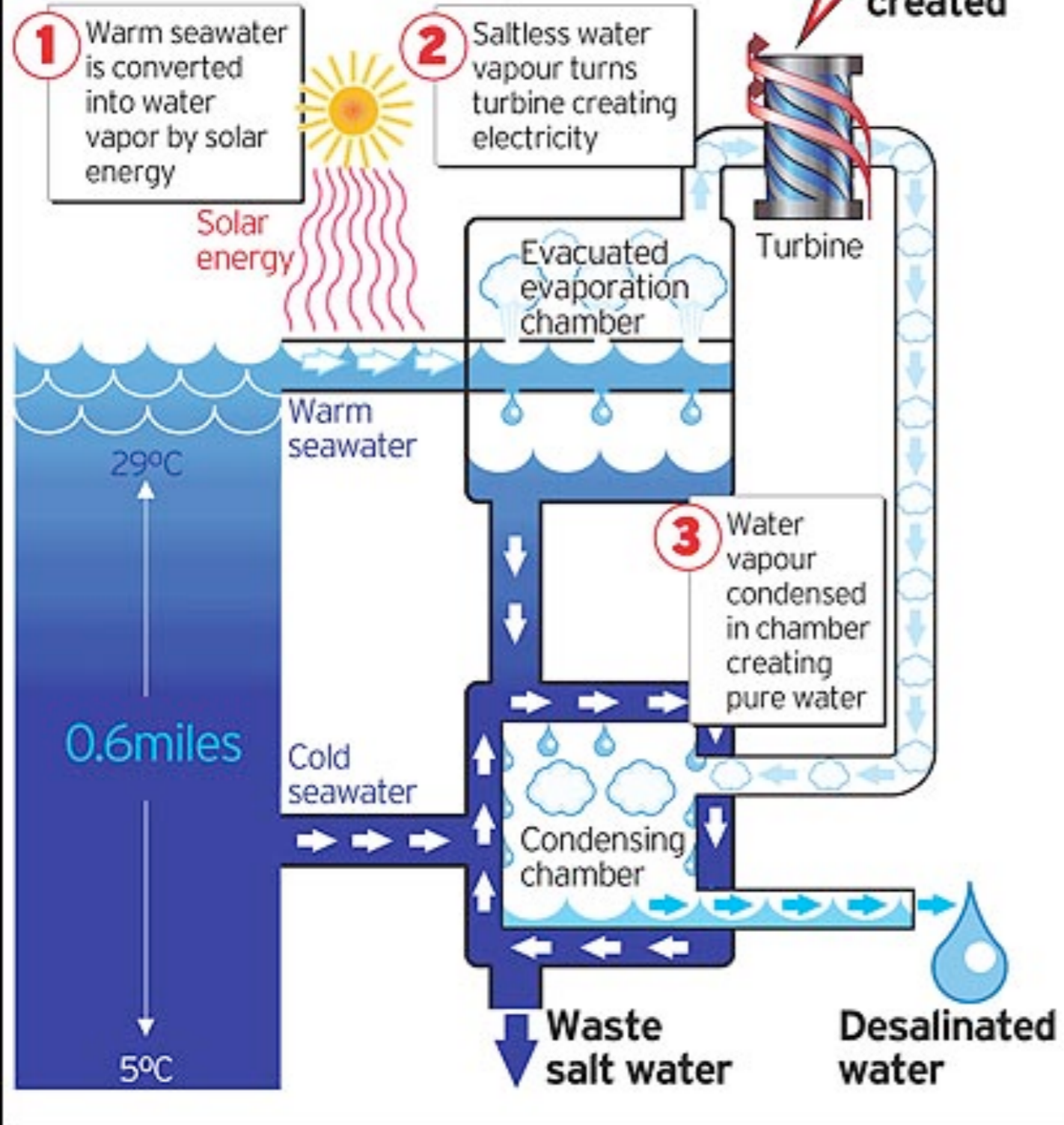
Ironically it is also the same phenomenon that allows schools of fish to swim as fast as they do. Now Dr. Michael M. Bernitsas and researchers at the University of Michigan are turning this 'threat' into a resource. Rather than suppressing VIV, Vivace actually creates and then harvests energy from VIV, and it does it all using slow water currents, a previously untapped source of sustainable energy.

OTEC
Ocean Thermal Energy Conversion

Water at surface is much warmer than water at 1000m depth. Can we use the temperature gradient of 20°C?



How ocean power operates



$$\eta = \frac{T_H - T_C}{T_H}$$

$$\eta = \frac{15}{300} \sim 5\%$$

If we cool 1000 gallons of water by 2°C, the power generated is 32 MW. At 5% efficiency this gives 1.6 MW output as usable power.

Offshore plants could produce Hydrogen that can be transported by ships..

Not a big player as yet, and rather cool response in US to this technology.

Lecture 16
Continued
May 24, 2011

Biofuels/Biomass

Motivation

All of a sudden, you know, we may be in the energy business by being able to grow grass on the ranch! And have it harvested and converted into energy! That's what's close to happening.

GWB: Feb 2006

Ford T was built to run on either ethanol or petrol!!
And then we discovered the middle eastern oil fields!

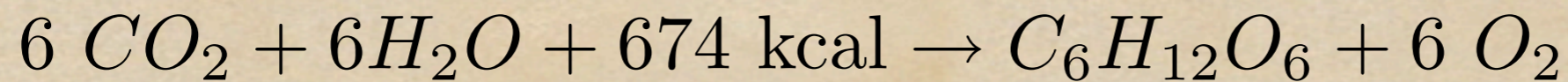
Photosynthesis

Biomass


- 5 billion yrs ago, the atmosphere had H_2 , He, N, CO_2 , NH_3 and water but no oxygen
- 3 billion years ago we had oxygen and plant life, H_2 , He escaped the earth.
- Photosynthesis is key, anaerobic processes (no Oxygen required) created carbohydrates

$C_x(H_2O)_y$ are carbohydrates, e.g. $C_6H_{12}O_6$ fructose

Light= ENERGY



Glucose



We may thus think of carbohydrate production by sunlight through this reaction.

A rough rule:

4300 cal energy needed to grow a gram of carb.

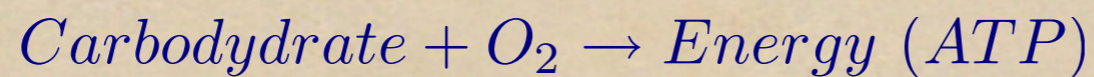
A mole is a gram molecular weight:

e.g. O_2 mole has 32 gms H_2 mole has 2 gms

CH_2O mole has 30 gms

(C=12, H=1, O=16 total 30)

Respiration: opposite of PS



Rate of carbohydrate production, agriculture, grains \rightarrow Alcohol \rightarrow Gasohol

We can say something quantitative about the total agricultural production on the basis of the solar constant!

$$\text{Solar constant} = 0.5 \text{ cal/min/cm}^2$$

47% reaches earth

$$\sim 500 \text{ cal/(cm}^2 \cdot \text{day)}$$

$\sim 500 \text{ cal}/(\text{cm}^2 \cdot \text{day})$

25% correct wavelength for PS

70% absorption by foliage

35% light useful for PS

$\sim 6\%$ of total

$\sim 30 \text{ cal}/(\text{cm}^2 \cdot \text{day})$

Convert cal to grams using
a rough rule:

4300 cal per gram of carb giving
 $75 \text{ gm}/(\text{m}^2 \times \text{day})$

Experimentally one finds about $70 \text{ gm}/(\text{m}^2 \cdot \text{day})$ of grain production averaged over many species

This comes out as $\sim 5\%$ of total energy, pretty close to our estimate of 6%!!

- Hubbert's data says total solar power available for PS is 40 TW.
- We can calculate the total potential production from this as $8 \times 10^{16} \text{ gm}/\text{year}$ on earth.

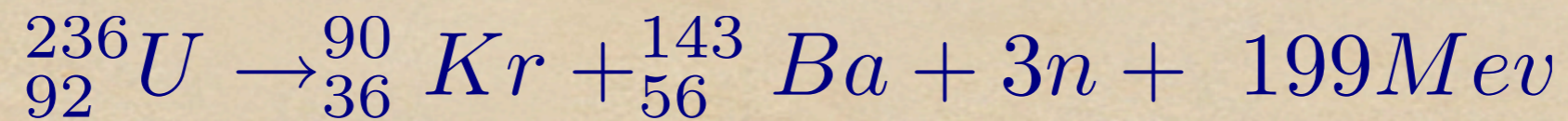
$$15 \text{ tons}/(\text{acre} \cdot \text{year}) \times 350 \times 10^6 \text{ acres} = 5.25 \times 10^9 \text{ Tons}/\text{year}$$
$$\text{at } 4300 \text{ (cal/gm)} \sim 79 \times 10^{15} \text{ BTU}$$

79 Quadrillion BTU versus 98 Q BTU used!!

Gasohol: 10% ethanol + petrol. Good for combustion efficiency and is promising.

Nuclear Energy

- Vast possibilities
- Much worry about safety, partly based on experience
- Further ideas for safer harvesting
- Need to know the basics:



Fission reaction: Need to understand the symbols and concepts.

General name	name	Charge	Mass	Mass $\times c^2$ <i>$E = m c^2$</i>
Nucleon Strongly interaction (Hadrons)	proton	+e	1.007 u	983 Mev
	neutron	0	1.008 u	984 Mev
Leptons	electron	-e	.00054 u	
	neutrino	0	~small	

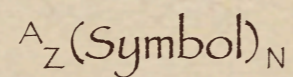
$$c = 3 \times 10^8 \text{ m/sec}$$

$$u = 1.66 \times 10^{-27} \text{ kg}$$

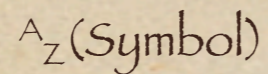
$$1 \text{ MeV} = 10^6 \text{ eV} = 1.6 \times 10^{-13} \text{ J}$$

Nucleus

A nucleus consists of Z protons and N neutrons.
Its mass is close to (but not exactly) (A+Z) u.
Their nomenclature is as follows:



or sometimes simply as



Nomenclature:

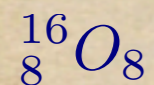
A= Mass number

Z= Atomic (or proton) number

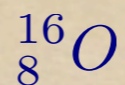
N= Neutron number

(Thanks for correction in
class today)

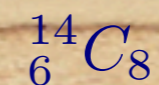
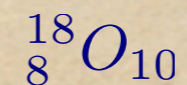
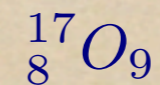
Example of abundant oxygen



or more simply



In nature we also find other
“isotopes” of Oxygen



A few important nuclei, and their isotopes

Hydrogen ${}^1_1\text{H}$

Stable hydrogen

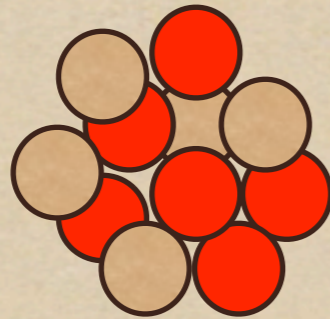
Deuterium ${}^2_1\text{H}$

Stable heavy hydrogen

Tritium ${}^3_1\text{H}$ *half-life* = 12 years

Helium	${}^4_2\text{He}$ ${}^3_2\text{He}$	
Carbon	${}^{12}_6\text{C}$ ${}^{14}_6\text{C}$	<p style="text-align: center;">-</p> <p style="text-align: center;">5600 yrs</p>
Uranium	${}^{238}_{92}\text{U}$ ${}^{235}_{92}\text{U}$	
Plutonium	${}^{244}_{94}\text{Pu}$ ${}^{239}_{94}\text{Pu}$	

Radius of a nucleus $\sim 10^{-15}$ m, i.e. a fermi



Strong interaction forces bind the nucleons together, overcoming their Coulomb repulsion by an even stronger attraction.

Binding energy and mass defect.

The reason a nucleus is stable is due to the binding energy. We can say:

$$E_{\text{nucleus}} \approx E_{\text{nucleons}} - E_{\text{Binding}}$$

or

$$E_{\text{Binding}} \approx E_{\text{nucleons}} - E_{\text{nucleus}}$$

$$M_{\text{defect}} = E_{\text{Binding}} / c^2$$